

Polyurea Definition

- A polyurea coating / elastomer is that derived from the reaction product of an isocyanate component and a resin blend component. The isocyanate can be aromatic or aliphatic in nature. It can be monomer, polymer, or any variant reaction of isocyanates, quasi-prepolymer or a prepolymer. The prepolymer, or quasi-prepolymer, can be made of an amineterminated polymer resin, or a hydroxyl-terminated polymer resin.
- The resin blend must be made up of amine-terminated polymer resins, and/or amine-terminated chain extenders. The amine-terminated polymer resins will not have any intentional hydroxyl moieties. Any hydroxyls are the result of incomplete conversion to the amine-terminated polymer resins. The resin blend may also contain additives, or non-primary components. These additives may contain hydroxyls, such as predispersed pigments in a polyol carrier. Normally, the resin blend will not contain a catalyst(s).

As per the Polyurea Development Association, 2000

Polyurea Definition

From the PDA Definition:

- * Two-part system
- *One part is Isocyanate component
 - •reactive isocyanate groups (-NCO)
- * Other part is a Resin blend component
 - reactive group is amine (primary or secondary)
 - (-NH₂ or -NH-), no hydroxyls (-OH)
- Reaction of the two components yields a urea linkage



Polyurea Definition

Other reactions for "urea" formation:

- Moisture cured urethane systems (low NCO content)
 - water / moisture reacts with isocyanate group
 - forms unstable carbamate, dissociates to amine group
- - oxime and phenol blocked isocyanates
 - · heat usually removes blocking agent, reaction then with amine
 - blocked amines (aldimines)
 - moisture converts aldimine to amine, reaction then with isocyanate





Polyurea Technology

- ★ Introduced in 1987 in USA
- ★ Volume usage in 1990, < 1,000 gallons
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 - < 3785 liters
- ***** 2005 Volume, 18 − 20 million gallons
 - 62 76 million liters
 - This is Worldwide!





From Polyurea Development Association (2003), Skeist Report (2004)

Polyurea Advantages

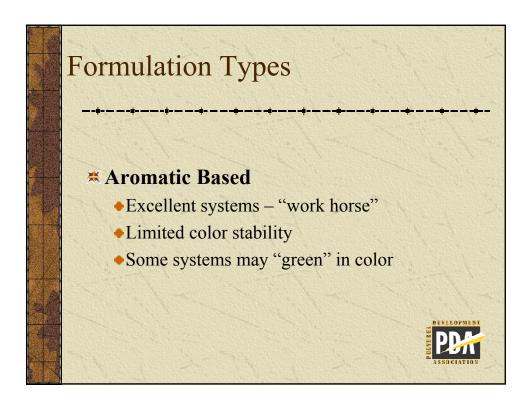
- * Fast reactivity, set and cure: rapid return to service
- ★ Insensitivity to ambient temperature
- Contains no Solvents, no VOC's & little to no odor
- **Excellent mechanical properties**

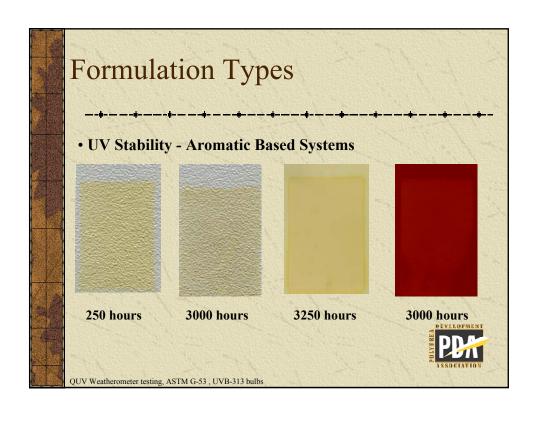


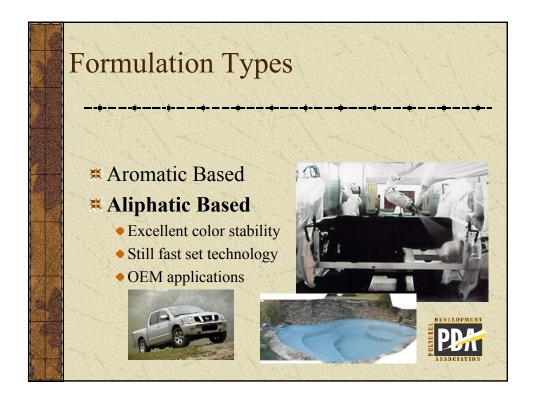
Polyurea Usage

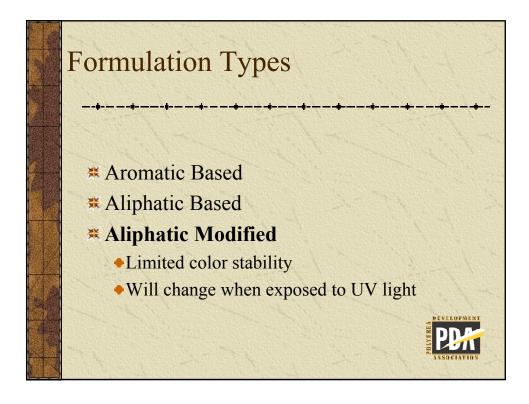
- ★ Concrete coatings and linings
 - Water & wastewater, chemical, water proofing
- * Steel coatings
 - Corrosion / abrasion protection, tank lining
- Molding work











Formulation Types

- * Aromatic Based
- * Aliphatic Based
- * Aliphatic Modified
- *** Polyaspartic Ester System**
 - Thin film application
 - •Slow set
 - Excellent color stability
 - Not as flexible





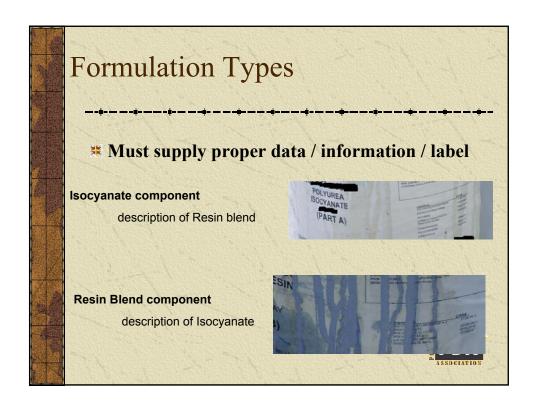
Formulation Types

High Temperature Glass Transition

System Type	T _g , °C
Aromatic based	230 - 260
Aliphatic based	60 - 80
Polyaspartic ester	60 - 80

Low temperature T_g, °C is about -40 to -50°C





** Polyurea is a thermoset material, not a thermoplastic ** As with all thermoset materials, there is some "shrinkage" associated with cure

Shrinkage Characteristics

- ★ Faster set, harder systems shrink more than
 the slower set, higher elongation systems
- ★ Formulation and processing dependant
- ★ Linear shrinkage values noted:
 0.5 up to 5% linear



Set and Cure

- ★ Polyurea technology noted as "fast cure"
- * Really a fast set technology
- ★ Overall cure faster than other technologies
 - Epoxy, polyurethane, vinyl ester

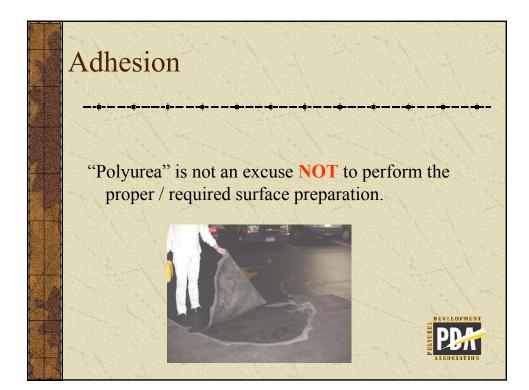


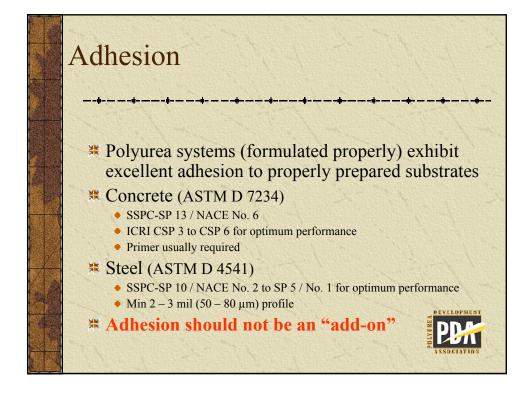
Set and Cure

- **Cure Time:** The period of time when a polyurea material attains it's ultimate physical properties.
- within a short period of time that would allow use in a particular application, but have **NOT** reached ultimate physical properties. The cure time varies on different formulations, with the high hardness, low elongation systems reaching full cure faster that the slower, higher elongation systems. This ultimate "cure" may be 24 to 48 hours plus after application.

Set and Cure

- ★ Faster set / gel system may not properly wet the substrate for adhesion
- Faster set systems may also have issue with overspray
- ** When can you return to service / expose to traffic?

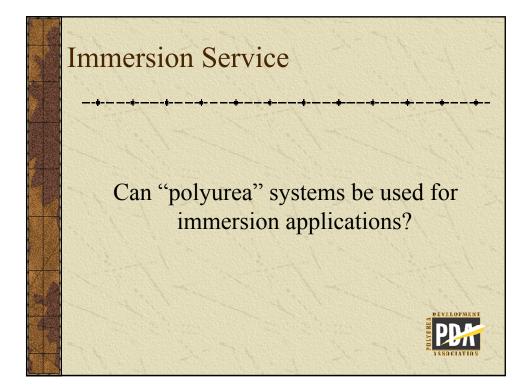


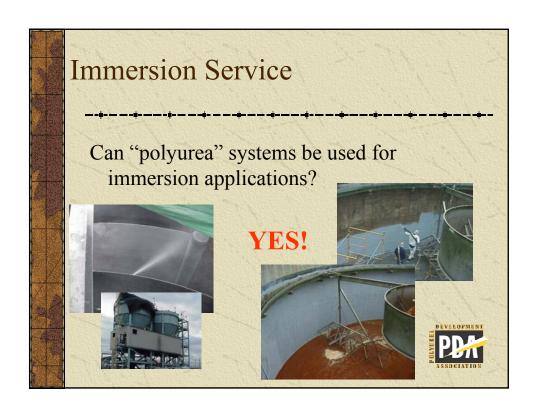


Primers ______

- * Typically required to enhance adhesion
 - Porous substrates
 - Reduce or eliminate pinholing
- Epoxy or polyurethane based
 - •Should be matched to the polyurea system
- *Only as good as what you are sticking to!









Immersion Service

- * Important criteria:
 - Properly formulated system
 - Moisture Vapor Transmission understanding
 - MUST be pinhole free
 - Especially for wastewater applications
 - MUST be processed properly





Chemical Resistance

- Similar to polyurethanes polyether backbone
- ★ More resistant to hydrolysis (urea linkage)
- Better alkali resistance (high pH)
- Exposure temperature significantly affects performance
- Some formulation subject to oxidation
- "Modified" systems showing improved performance



Application Equipment

- ** Plural component spray application
 - Capable of high, continuous flow
 - Heating is required
- ***** Equipment setup to application work
 - Downsize out to spray gun
 - Must have positive feed to unit















Applicator Training

- Offered by Polyurea System Supplier
 - Capable of high, continuous flow
 - Specific to their "polyurea" products
- ** Polyurea Development Association
 - Polyurea Spray Applicator Course
 - First course of it's kind available to industry
 - Extensive and "hands on", 4-day course
- * Other Associations; SSPC and NACE



Conclusion

- **Support** the product
 - Performance data / testing
- * Choose the proper system for the application
- # Have application history
- Training programs available
- * Working on new developments
- **✷** Ask Questions

